

GC Sequence Specific Recognition by an *N*-Formamido, C-Terminus-Modified and Imidazole-Containing Analogue of Netropsin

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This communication describes, to our knowledge, the first example of a lexitropsin that recognizes solely GC sequences of DNA, and it will have immediate application in the development of new pharmaceutical agents,¹ artificial restriction enzymes,² as well as DNA probes.³ There is a current interest in the design of small molecules that can bind to the minor groove of DNA, especially to GC-rich sequences. This interest stems from the observation that most minor groove binders are AT sequence specific⁴ and that GC-rich sequences are commonly found in genomes of mammals, including humans. A functional role of such GC-rich sequences is suggested by their frequent occurrence in genes associated with proliferation, including a number of oncogenes.⁵

Netropsin **1** (see Figure 1) and distamycin, members of a class of naturally occurring antibiotic oligopeptides that bind to the minor groove of (A/T)₄ and (A/T)₅ sequences, respectively,^{6,7} have been widely employed as models for the design of GC sequence-selective lexitropsins.⁸ It was shown that replacement of the pyrrole moieties of netropsin with imidazoles⁸⁻¹⁰ and removal of one of the positive charges, such as in analogue **2**,¹⁰ permitted GC sequence recognition. Due to steric interactions between the methylene groups of **2** with the guanine-2-NH₂ group in the minor groove of a GC base pair, the C-terminus of this ligand was forced to bind to an AT base pair, thus explaining its selectivity for the sequence 5'-(G/C)₃(A/T).¹⁰ It was also demonstrated that a truncated netropsin analogue, **3**, selectively binds to the sequence 5'-ATTG-3' when challenged with d(CGCAATTGCG)₂, indicating that the modified C-terminus could accept a GC base pair.¹¹ While the design of GC sequence-selective lexitropsins,^{9b,c} such as **2**,¹⁰ was viewed as a significant achievement, Kopka and Larsen stated that the development of netropsin analogues which can recognize pure GC sequences still remains a formidable task.^{8c}

In this communication we report the DNA binding studies of an imidazole-containing analogue, **4**¹² (Figure 1). The N- and C-termini of this compound contain a formamido and a carboxamido moiety, respectively, replacing the guanidiniumylacetyl and propanamidinium groups of netropsin, and it represents a departure from our previous prototype compounds such as **2**. The inclusion of a basic (dimethylamino)ethyl moiety (pK_a ~9.8) at the -terminal imidazole of **4** ensured that it would be protonated at a physiological pH of 7.4, thus making it monocationic. On the basis of recent reports

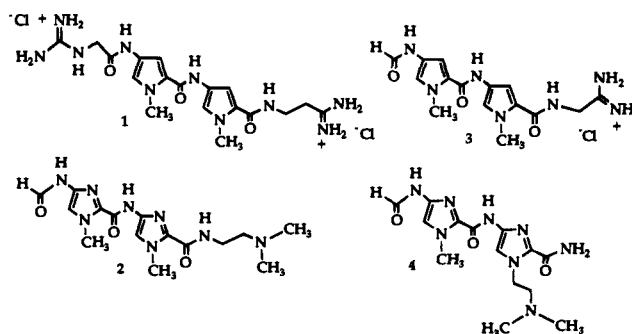


Figure 1. Structures of netropsin **1**, the monocationic imidazole-containing lexitropsin **2**, the truncated netropsin analogue **3**, and the target compound **4**.

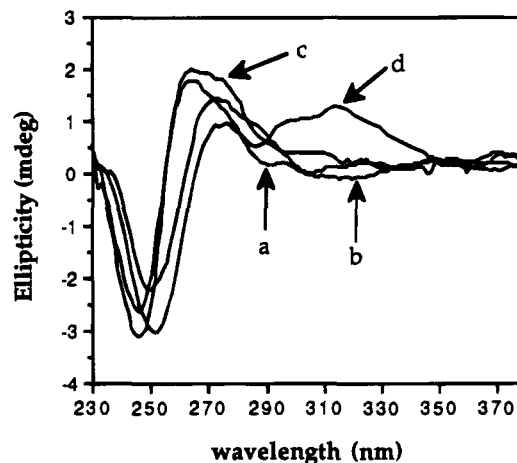


Figure 2. Representative overlaid CD spectra of the titration of compound **4** to poly(dA-dT) and poly(dG-dC). Spectra a and b are for free poly(dA-dT) and poly(dG-dC), respectively. Spectra c and d represent the titration of compound **4** to the respective DNAs at *r'* values of 0.8.

on the DNA binding properties of *N*-pyrrole polyamine-substituted distamycin analogues, the positively charged dimethylammonium group of agent **4** should be positioned to electrostatically interact with the phosphate backbone thereby stabilizing the ligand:DNA complex.¹³

The apparent binding constants of analogue **4** to calf thymus DNA, T4 coliphage DNA, poly(dG-dG) and poly(dA-dT) determined using an ethidium displacement assay¹⁴ were $(1.37 \pm 0.05) \times 10^5$, $(1.9 \pm 0.3) \times 10^4$, $(1.05 \pm 0.04) \times 10^5$, and $(6.5 \pm 0.5) \times 10^4$ M⁻¹, respectively. According to these data, analogue **4** is a weaker binder than netropsin **1** which has a binding constant to calf thymus DNA of 1.87×10^6 M⁻¹ under identical conditions.^{9a} The difference may result from several factors, including the smaller number of positively charged groups and van der Waals sites on **4** compared to those on **1**. The results, however, show that compound **4** binds approximately twice as strongly to poly(dG-dC) than to poly(dA-dT). In comparison, netropsin binds approximately 1000-fold more strongly to poly(dA-dT) than to poly(dG-dC),¹⁵ thus demonstrating the acceptance of GC base pairs by compound **4**. The interaction of **4** to T4 coliphage DNA suggests that it interacts in the minor groove of DNA.^{16a,b}

The binding of compound **4** to poly(dA-dT) and poly(dG-dC) was also studied by CD. The experiments were identically performed with regard to temperature and concentrations of drug and DNA at defined *r'* values, the ratio of the moles of drug to DNA base pairs.

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constants of compound **4** to those of **2**, a known minor groove binder.¹⁰ The binding of agent **4** to the minor groove implies that the amido protons should be able to form bifurcated hydrogen bonds with guanine-N3 and cytosine-O2 in the floor of the minor groove to stabilize the ligand:DNA complex. In addition, the strict recognition of pure GC sequences also suggests that the imidazole-N3 atoms on the concave face of compound **4** could provide both the space for accommodating and hydrogen bonding with the exocyclic guanine-2-NH₂ group. As a control in the footprinting experiment, compound **2** (Figure 3a, lane c) produced strong footprints at other sites on the same fragment of DNA that contain the expected 5'-(G/C)₃(A/T)-3' sequence.¹⁰ Further studies on interactions of compound **4** with oligodeoxyribonucleotides by ¹H-NMR and X-ray studies are in progress, and the results will be reported in due course.

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Supplementary Material Available: The synthetic scheme for the preparation of compound **4**, its ¹H-NMR spectrum as well as the densitometric scans of the MPE footprinting experiments are given (4 pages). Ordering information is given on any current masthead page.

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- (12) The synthetic scheme for the preparation of compound **4** is given in the supplementary material section: mp 170 °C dec; TLC (10% MeOH/CHCl₃) *R_f* 0.77; ¹H-NMR (CDCl₃) δ 9.18 (br s 1H), 8.36 (s, 1H), 7.60 (br s, 1H), 7.55 (s, 1H), 7.46 (s, 1H), 7.05 (br s, 1H), 5.29 (br s, 1H), 4.57 (t, 7.2, 2H), 4.08 (s, 3H), 2.73 (t, 7.8, 2H), 2.30 (s, 6H); IR (CHCl₃ cast) ν 3412, 2921, 1665, 1534, 1458, 1376 cm⁻¹; UV (water) λ_{max} 205, 302 nm (ε = 104 950, 35 019 M⁻¹ cm⁻¹); MS (FAB, NBA/TFA) *m/z* (intensity) 349 (34, M + H); FAB-HRMS *m/z* 349.1804 (C₁₄H₂₀N₈O₃ + H), calcd 349.1787.
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